No. of Printed Pages: 4 Q.8 Define the term Antidotes. Roll No. 2112 Q.9 Why Nitric acid is added in the limit test of **Pharmacy** chloride? **Subject: Pharmaceutical Chemistry - 1** Q.10 How do Astringents Act? Time: 3 Hrs. M.M.: 80 **SECTION-A SECTION-B** Note: Objective type questions. All questions are **Note:** Very short answer type questions. Attempt any (10x1=10)compulsory five questions out of seven questions. 5x3=15Write the General Formula of Borax. Q.11 Mention the Classification of Non Systemic Define Dentifrices. Antacids. Mention the Antacid salt which may cause Q.12 Mention the uses of Potassium Permangnate. constipation. Q.13 Mention the classification of Antidotes. Mention the use of Zinc Oxide. Q.14 Mention role of extracellular ion Sodium. Define the term Achlorhydria. Q.15 Mention uses of Zinc Chloride. Define the term emetics. Q.16 Mention the identification test for Bicarbonate Mention two extracellular ions used as Q.7 (Anion) Electrolytes. (1) (2)2112 2112

Q.17 Write a brief note on Buffers.

SECTION-C

- **Note:**Restricted response questions of 5 marks each out of which 5 are to be attempted. 5x5=25
- Q.18 Write a note on preparation, Properties and uses of Hydrogen Peroxide.
- Q.19 Write a brief note on Classification and uses of electrolytes.
- Q.20 Write a note on preparation, properties and uses of Nitrous Oxide.
- Q.21 Write a brief note on Expectorants.
- Q.22 Write a brief note on official compound of iron.
- Q.23 Mention the principle of limit test of chloride.
- Q.24 Write a note on preparation, properties and uses of Aluminium Hydroxide Gel.

SECTION-D

- **Note:** Extended response or Essay type questions of 10 Marks each out of which 3 questions to be attempted. 3x10=30
- Q.25 Write a detail note on sources of impurities in Pharmaceuticals.
- Q.26 Explain the limit test of arsenic with Diagram, principle and procedure.
- Q.27 Write a detail note on calcium containing Antacids.
- Q.28 Define the term Radiopharmaceuticals Mention their Biological effects and applications.

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